## A NOVEL CAGE COMPOUND CONTAINING TWO PHOSPHORUS ATOMS

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A diphospha analogue of barrelene, 2,3,5,6,7,8-hexakis(trifluoromethyl)-1,4diphosphabicyclo[2.2.2]octa-2,5,7-triene (I)<sup>1)</sup>, had been reported to be stable to atmospheric oxygen or bromine dissolved in chloroform, because of steric hindrance of the trifluoromethyl groups. But we reported<sup>2)</sup> that methanol added to a double bond of I in the presence of RhCl<sub>3</sub> and that 2,3,5,6-tetrakis(trifluoromethyl)-1,4-diphosphabenzene was derived from this adduct by thermolysis.

In this paper, we wish to report the Diels-Alder reaction of I with cyclobutadiene and a derivation to a new cage compound from the adduct.

Although compound I did not react with furan or tetraphenylcyclopentadienone, a less bulky diene was expected to react with I. Therefore, we investigated the reaction of I with cyclobutadiene. In fact, I reacted with cyclobutadiene, generated from cyclobutadieneiron tricarbonyl, at room temperature in acetone to give two isomeric 1:1 adducts (II and III). II: 33.2%; mp 111-112° (colorless needles from chloroform); ir (KBr) 1600 cm<sup>-1</sup> (C=C); <sup>1</sup>H-nmr (CDCl<sub>3</sub>)  $\boldsymbol{\delta}$  3.95 (2H, b,  $\boldsymbol{\xi}$ C-H), 6.00 (2H, b, C=C-H); <sup>19</sup>F-nmr (CDCl<sub>3</sub>) $\boldsymbol{\delta}$  <sup>3</sup>) -2.84 (6F, d, J<sub>PF</sub>=41.7 Hz), -6.04 (6F, d, J<sub>PF</sub>=39.5 Hz), -6.08 (6F, d, J<sub>PF</sub>=39.5 Hz); mass spectrum m/e 600 (M<sup>+</sup>). III: 23.7%; mp 97-98° (colorless needles from chloroform); ir (KBr) 1600 cm<sup>-1</sup> (C=C); <sup>1</sup>H-nmr (CDCl<sub>3</sub>) $\boldsymbol{\delta}$  3.00 (2H, m,  $\boldsymbol{\xi}$ C-H), 6.22 (2H, s, C=C-H); <sup>19</sup>F-nmr (CDCl<sub>3</sub>) $\boldsymbol{\delta}$  -5.40 (6F, d, J<sub>PF</sub>=40.6 Hz); -6.06 (6F, d, J<sub>PF</sub>=38.4 Hz), -6.16 (6F, d, J<sub>PF</sub>=33.8 Hz); mass spectrum m/e 600 (M<sup>+</sup>).

Next, we examined the photolysis of II and III to establish the configuration of the two isomers. Irradiation of II in acetone gave 2,5,7,10,11,12hexakis(trifluoromethy1)-1,6-diphosphahexacyclo[4.4.2.0<sup>2,5</sup>.0<sup>3,9</sup>.0<sup>4,8</sup>.0<sup>7,10</sup>]dodecene-11 (IV) in 74.3% but that of III did not. This fact shows that II is an endo isomer and that III is an exo one. IV: mp 201-203°(white plates from acetone); ir (KBr) 1610 cm<sup>-1</sup> (C=C); <sup>1</sup>H-nmr (CD<sub>3</sub>COCD<sub>3</sub>) $\delta$  3.96 (4H, s,  $\gtrsim$ C-H); <sup>19</sup>F-nmr (CD<sub>3</sub>COCD<sub>3</sub>) $\delta$ -1.06 (12F, d, J<sub>PF</sub>=41.3 Hz), -4.80 (6F, d, J<sub>PF</sub>=38.6 Hz); mass spectrum m/e 600 (M<sup>+</sup>). All these data support the structure of IV, which is an interesting compound of a new ring system containing two phosphorus atoms.



## REFERENCES

- 1) C.G. Krespan, J. Amer. Chem. Soc., 83, 3432 (1961).
- 2) Y. Kobayashi, I. Kumadaki, A. Ohsawa, H. Hamana, Tetrahedron Lett., in press.
- 3) Benzotrifluoride:+0 ppm as an internal standard.